

### Abstract

To provide perspective on current and proposed occupational exposure limits (OELs) for peracetic acid (PAA; CAS 79-21-0) we evaluated PAA toxicity with the aim of understanding uncertainties and their implications for the resulting OEL. The database for PAA is limited and no single study is definitive. Two unpublished reports on human exposures to PAA provide some concentration-response data, indicating that a sensitive acute effect of PAA exposure is eye and respiratory tract irritation, but the studies differ quantitatively. These differences are not surprising, in light of the differences in exposures (apparently pure PAA vapor vs. an aerosol of a mixture), the subjective nature of the reporting, and the likely small sample sizes. The studies are also limited by the lack of clear concentration-duration-response data. Nonetheless, the studies provide a reasonable estimate of the threshold for the onset of irritation in humans in the range of 0.53 mg/m<sup>3</sup> for up to 3 hours and 1.56 mg/m<sup>3</sup> for up to 45 minutes. RD<sub>50</sub> (concentration estimated to cause a 50% depression in respiratory rate) data in mice and rats provide additional information on the irritant potency of PAA. The RD<sub>50</sub> in mice was 17 mg/m<sup>3</sup> for pure PAA vapor and 12 mg/m<sup>3</sup> for a commercial mixture. The rat  $RD_{50}$  was 21.5 to 24.1 mg/m<sup>3</sup>. Based on the array of human data and the RD50 values in rodents, we calculated potential TWA OELs ranging from 0.26 to 1.56 mg/m<sup>3</sup>. A similar range of  $0.62 - 2 \text{ mg/m}^3$  is found among the published OELs, and any of these values could be justified as protective of worker health given the uncertainties in the data and the precision of the OEL methodology. More definitive sensory irritation studies would further clarify selection of a value in this range. Given the extant data, the ultimate OEL choice within a range of reasonable values is a policy-based risk management decision, not a scientific one. The optimal time averaging approach is also not clearly established by the data; however, a combination of a TWA with a STEL is the recommended risk management option.

### Background

- Produced from the acid-catalyzed reaction between acetic acid and hydrogen peroxide (Sanchez and Meyers, 2000)
- Technical grades of peracetic acid contain approximately 40% PAA in acetic acid with residual hydrogen peroxide (ACGIH, 2013)
- Used commercially as a chemical intermediate, bleaching agent, and sterilant, as well as in the formation of epoxides, epoxy resins, and the bleaching of textiles
- Potent oxidizing chemistry that is highly corrosive. Lower concentrations are irritating to eyes and mucous membranes
- Vapor pressure of 14.5 mmHg at 25°C. Saturated vapor concentration of 59,300 mg/m<sup>3</sup> (ACGIH, 2013)
- Pungent odor with no clearly established threshold. Odor can serve as a warning property for the presence of atmospheric PAA
- Atmospheric monitoring of PAA can be problematic, particularly in atmospheres with appreciable levels of chlorine and/or hydrogen peroxide.

## <u>Time Averaging Approach - TWA,</u> <u>Ceiling or STEL</u>

- Ceiling limit often established for rapidly acting sensory irritants that lack significant secondary or chronic tissue toxicity - Ceiling limit should protect from not only sensory irritation, but also subclinical cytotoxic responses that could accumulate and yield other longer-term effects.
  - PAA is an irritant that can also cause cytotoxicity. Insufficient dose-response information is available to ensure that the absence of sensory response also protects from accumulating subclinical cytotoxic responses.
- Thus, a ceiling limit approach is not recommended.
- A short-term exposure limit (STEL) can protect protection from rapid onset and transient effects such as sensory irritation.
- The STEL is typically accompanied by a TWA derived separately to ensure protection from longer-term effects.
- If the upper respiratory tract (URT) irritation is the only adverse effect (i.e., tissue damage does not occur), it can be appropriate to establish a STEL in the absence of a longer-term TWA.
- This is sound physiologically, but has practical problems.
- In practice, the STEL is a 15-minute TWA that reflects a concentration measured over a 15-minute period. In the absence of a longer-term (e.g., 8-hour) TWA-based limit, control at the STEL is equivalent to a series of consecutive 15 minute-TWAs, and thus equivalent to a longer-term TWA.
- Usually, where a longer-term TWA is established along with a STEL, the TWA includes an implied limit to the number and duration of excursions above the TWA and below the STEL, which means that the STEL does not become equal to the longer-term TWA value.
- For PAA, the data for longer-term effects are limited and the MOA suggests some cytotoxic potential. Therefore a limit based solely on the STEL is not recommended.

# Approaches for Deriving an OEL for Peracetic Acid and Occupational Risk Management

Nathan Pechacek<sup>1</sup>, Andrew Maier<sup>2</sup>, Lynne T. Haber<sup>2</sup> <sup>1</sup>ECOLAB, St. Paul, Minnesota, USA <sup>2</sup>TERA, Cincinnati, Ohio, USA

Table I. Key Pera	acetic A	Acia worker E	xposure Stud	ies and Elle	ct Levels	Table 3. Summary of Potential and Existing Exposure Limits					
		Concent	ration			Value	Basis	Point of Departure	Factor	Final Value	
Duration		ppm	mg/m <sup>3</sup>	Ef	fect	Derived here	McDonagh	$0.53 - 1.56 \text{ mg/m}^3$	1-3 (midpoint is	$0.26 - 1.56 \text{ mg/m}^3$	
		McDonagh	(1997) (vapor)				(1997)	(vapor)	2)	(0.09 – 0.5 ppm)	
Not clear, but poss bly as long as 3 ho	out possi- as 3 hours 0.5-0.6 1.56-1.87 Not consider 1.56-1.87 have been c "unpleasant tended perior		red immedi- g but would onsidered for an ex-	Derived here	Fraser and Thorbinson (1986)	$\frac{1.56 \text{ mg/m}^3}{(\text{aerosol})}$	1-3 (midpoint is 2)	$0.56 - 1.56 \text{ mg/m}^3$ $0.4 - 0.57 \text{ mg/m}^3$			
Not clear, but pose bly as long as 3 ho	si- ours 0.1	13-0.17	0.40-0.53	Tolerable, no	of time of unpleasant	Derived here	$RD_{50}$	(3.4 - 5.4  ppm) (vapor) 21.5 to 24.1 mg/m <sup>3</sup>	30 (Schaper, 1993)	$(0.13 - 0.37 \text{ mg/m}^3)$ (0.13 - 0.18 ppm) based on the mouse vapor dat or 0.71-0.8 mg/m <sup>3</sup> base	
Not clear, but pose bly as long as 3 ho	si- ours 0.1	17	0.53 No lacrimati		on			$\frac{(\text{aerosol})}{1.56 \text{ mg/m}^3 (0.5)}$		on the rat aerosol data	
Fraser and Thorbinson (1986) (aerosol)							McDonagh	ppm)	3 – Human vari-		
5-10 minutes	N,	/A - aerosol	6.23	Extreme disc mucous men	scomfort of mbranes (NRC AEGL-1 10 min – 8 hour (NRC, 2010)		(1997); Fraser and Thorbin- son (1986)	tating [McDonagh (1997)]; No discom-	Low variability based on mode	0.52 mg/m <sup>3</sup> (0.17 ppm)	
15-20 minutes	$\mathbf{N}_{i}$	/A - aerosol	3.12-4.67	Discomfort of membranes	of mucous			Thorbinson (1986)]	or action (MOA)		
25 minutes	N,	/A - aerosol	3.12	Discomfort	tolerable	$^{2}$ NRC AEGL-2 10 min – 8 hour	Fraser and Thorbinson	4.7 mg/m <sup>3</sup> (1.5 ppm) Slight to tolerable discomfort for expo-	3 – Human vari- ability Low variability	1.56 mg/m <sup>3</sup> (0.5 ppm)	
30 minutes	$ \mathbf{N}\rangle$	/A - aerosol	1.56-3.12	Discomfort	mild	(INRC, 2010)	(1986)	20 minutes	based on MOA		
35-45 minutes	-45 minutes $N/A$ - aerosol $\leq 1.56$ No discomfort			ort	NIC TLV-STEL	McDonagh (1997); Fraser and Thorbinson (1986); $RD_{50}$ $\begin{pmatrix} 1.24 \text{ mg/m}^3 \\ (0.4 \text{ mpm}) \end{pmatrix}$			$1.24 \text{ mg/m}^3$		
Table 2. Key Per Study	acetic A	Acid Data from	n Animal Exp	Duration	RD50	15-minute STEL (Gagnaire et al., 2002) TLV-TWA	Based on 0.1 x	$x RD_{50} \text{ of } 0.54 \text{ ppm}$		1.56 mg/m <sup>3</sup> (0.5 ppm) 0.62 mg/m <sup>3</sup>	
Gagnaire et al. (2002, as cited by ACGIH, 2011; OECD 2008	louse	<ul> <li>Pure peracetic acid vapor evaporated from a buffered (pH 7) commercial mixture containing 39% peracetic acid, 45% acetic acid and 6% hydrogen peroxide</li> <li>Peracetic acid vapor from a commercial mixture contain- ing 36% peracetic acid, 53% acetic acid and 11% hydro- gen peroxide</li> </ul>		5.4 ppm (17 mg/m³)	(Gagnaire et al., 2002) ECHA Worker DNEL, local ef-	Based on 0.3 x RD <sub>50</sub> of 0.54 ppm (0.2 ppm)		(0.2 ppm)			
Gagnaire et al. M (2002)	Iouse			3.8 ppm (12 mg/m <sup>3</sup> )	fects – acute/ short term or long-term expo- sure			0.6 mg/m <sup>3</sup> (0.2 ppm)			
Janssen (1989c, as cited by OECD, 2008; ta NRC, 2010)	Iale Wis ar Rat	Nis-Aerosol generated from a mixture of 15% peracetic acid, 14% hydrogen peroxide, utes25 min- utes			21.5 to 24.1 mg/m <sup>3</sup> (no clear ex- posure- response trend)	Various MAKs Long-term value (European Commission - EC, 2000)	Documentation not available			1 to 1.4 mg/m <sup>3</sup> (0.32 to 0.45 ppm)	
The only repeated exposure studies were conducted at 150 mg/m <sup>3</sup> and higher, and caused slight to evere squamous metaplasis of the nasal turbinates in all exposed groups.					Various MAKs 30-minute value (EC, 2000)	Documentation not available			2 mg/m <sup>3</sup> (0.64 ppm)		

		Concentration				Value	Basis	Point of Departure	Uncertainty Factor	Final Value
Duration		ppm	mg/m <sup>3</sup>	Ef	fect	Derived bore	McDonagh	$0.53 - 1.56 \text{ mg/m}^3$	1-3 (midpoint is	$0.26 - 1.56 \text{ mg/m}^3$
	Μ	cDonagh (	(1997) (vapor)			Derived here	(1997)	(0.17 - 0.5  ppm) (vapor)	2)	(0.09 – 0.5 ppm)
Not clear, but possi- oly as long as 3 hours	0.5-0.6	5	1.56-1.87	Not consider ately irritatin have been co "unpleasant	red immedi- g but would onsidered for an ex-	Derived here	Fraser and Thorbinson (1986)	1.56 mg/m <sup>3</sup> (aerosol) $12.17 mg/m^3$	1-3 (midpoint is 2)	$0.56 - 1.56 \text{ mg/m}^3$
Not clear, but possi- bly as long as 3 hours	0.13-0	.17	0.40-0.53	tended perio Tolerable, no	d of time" ot unpleasant	Derived here	<b>RD</b> <sub>50</sub>	(3.4 - 5.4  ppm) (vapor) 21.5 to 24.1 mg/m <sup>3</sup>	30 (Schaper, 1993)	$(0.13 - 0.37 \text{ mg/m}^3)$ (0.13 - 0.18 ppm) based on the mouse vapor dat or 0.71-0.8 mg/m <sup>3</sup> based
bly as long as 3 hours	0.17		0.53	No lacrimati	on			(aerosol) $1.56 \text{ mg/m}^3 (0.5)$		on the rat aerosol data
F	raser ar	nd Thorbin	ison (1986) (a	erosol)		NRC AEGL-1	McDonagh (1997); Fraser and Thorbin- son (1986)	ppm) Not immediately irri-	<ul> <li>3 – Human vari-</li> <li>ability,</li> <li>Low variability</li> <li>based on mode</li> <li>of action (MOA)</li> </ul>	
5-10 minutes	N/A -	- aerosol	6.23	Extreme disc mucous men	comfort of hbranes	10 min – 8 hour (NRC, 2010)		tating [McDonagh (1997)]; No discom- fort [Fraser and		0.52 mg/m <sup>3</sup> (0.17 ppm)
15-20 minutes	N/A -	- aerosol	3.12-4.67	membranes	SI macous			Thorbinson (1986)]         4.7 mg/m <sup>3</sup> (1.5 ppm)		
25 minutes	N/A -	- aerosol	3.12	Discomfort	tolerable	<sup>2</sup> NRC AEGL-2 10 min – 8 hour (NRC, 2010)	Fraser and Thorbinson (1986)	Slight to tolerable discomfort for expo- sure durations up to	3 – Human vari- ability Low variability based on MOA	1.56 mg/m <sup>3</sup> (0.5 ppm)
	IN/A -	- aerosoi						20 minutes		
$5-45 \text{ minutes}$ N/A - aerosol $\leq 1.56$ No discomfort				NIC TLV-STEL (ACGIH, 2013) McDonagh (1997); Fraser and Thorbinson (1986)			oinson (1986); RD <sub>50</sub>	1.24 mg/m <sup>3</sup> (0.4 ppm)		
Table 2. Key Perace         Study       Specentiation	etic Acio	d Data from Form	n Animal Expound	Duration	RD50	15-minute STEL (Gagnaire et al., 2002) TLV-TWA (Gagnaire et al.,	Based on 0.1 x Based on 0.3 x	$3 RD_{50}$ of 0.54 ppm		1.56 mg/m <sup>3</sup> (0.5 ppm) 0.62 mg/m <sup>3</sup> (0.2 ppm)
Gagnaire et al. (2002, as cited by ACGIH, Mou 2011; OECD 2008	ev ev (p co ac hy	evaporated from a buffered (pH 7) commercial mixture containing 39% peracetic acid, 45% acetic acid and 6% hydrogen peroxide			5.4 ppm (17 mg/m <sup>3</sup> )	2002) ECHA Worker DNEL, local ef- fects – acute/	0.6 mg/m <sup>3</sup>			
Gagnaire et al. (2002)	Se in	Peracetic acid vapor from a commercial mixture contain- ing 36% peracetic acid, 53% 1 hour acetic acid and 11% hydro- gen peroxide		3.8 ppm (12 mg/m <sup>3</sup> )	short term or long-term expo- sure	Documentation not available		(0.2 ppm)		
Janssen (1989c, as cited by OECD, 2008; tar R NRC, 2010)	e Wis- m Lat ac ar	Aerosol generated from a - mixture of 15% peracetic acid, 14% hydrogen peroxide, utes and 28% acetic acid			21.5 to 24.1 mg/m <sup>3</sup> (no clear ex- posure- response trend)	Various MAKs Long-term value (European Commission - EC, 2000)	Documentation not available			1 to 1.4 mg/m <sup>3</sup> (0.32 to 0.45 ppm)
The only repeated exposure studies were conducted at 150 mg/m <sup>3</sup> and higher, and caused slight to severe squamous metaplasis of the nasal turbinates in all exposed groups.					Various MAKs 30-minute value (EC, 2000)	Documentation not available			2 mg/m <sup>3</sup> (0.64 ppm)	

### Abstract No. 2226 Poster Board No. 424





## Recommendations/Conclusions Regarding Time Averaging

- Based on MOA considerations, PAA may have longer-term effects for which a STEL approach alone does not provide adequate protection.
- The selection of a STEL without a TWA should not be driven by the absence of robust longer-term effects data.
- The use of a TWA with a STEL at a higher concentration reflects the different biological responses (sensory irritation and cytotoxicity) in ways that give the risk manager flexibility.
- The use of a TWA with a STEL may also be a better reflection of the temporal patterns in biology, since some elements of the irritation response are not completely independent of exposure duration.
- The use of a TWA and STEL together provides meaningful guidance for excursion limits and is most consistent with current practice.

### Overall Conclusions

- We calculated potential TWA OELs ranging from 0.26 to 1.56 mg/m<sup>3</sup>.
- This is similar to the range of  $0.62 2 \text{ mg/m}^3$  found among the published OELs
- Any value within the range could be justified as protective of worker health in light of the uncertainties in the data and the precision of the OEL methodology.
- More definitive sensory irritation studies would further clarify selection of a value in this range.
- The ultimate OEL choice is a policy-based risk management decision, not a scientific one.
- The optimal time averaging approach is not clearly established by the data; however, a combination of a TWA with a STEL is recommended as a preferred risk management option.

### References

ACGIH® (American Conference of Governmental Industrial Hygienists), 2013. Peracetic Acid. Notice of Intended Change (NIC). Draft Documentation of the Biological Exposure Indices. ACGIH® Worldwide. Cincinnati, Ohio.

EC (European Commission), 2000. Peracetic Acid (CAS No. 79-21-0). IUCLID Dataset. 2000 CD-ROM Ed. European Commission. European Chemical Bureau (online). Available at http://ecb.jrc.it/IUCLID-DataSheets/79210.pdf. Accessed July 20, 2012. Fraser, J.A.L., Thorbinson, A., 1986. Fogging Trials with Tenneco Organics Limited (30th June, 1986) at Collards Farm. Solvay Interox. Warrington, United Kingdom.

Janssen, P.J.M., 1989c. Acute Inhalation Study to Investigate the Respiratory Irritating Properties of Proxitane 1507 in Male Rats. Report No. S. 8912. Int. Doc. No. 56645/40/89.

McDonagh, J., 1997. Atmospheric Monitoring of Peracetic Acid on the Existing Caprolactone Plant Distillation Houses A and B, Assessment of Results. Document No. EE970192.M01. Memorandum to R.A. Haffenden et al. from J. McDonagh. Solvay Interox. Warrington, United Kingdom.

NRC (National Research Council), 2010. Acute Exposure Guideline Levels for Selected Airborne Chemicals: Volume 8. National Academies Press. Available at http://www.nap.edu/catalog/12770.html. Accessed July 20, 2012

OECD (Organisation for Economic Cooperation and Development), 2008. SIDS Initial Assessment Report for Peracetic Acid. SIAM 26. Paris, France.

Sanchez, J., Meyers, T.N., 2000. Peroxides and Peroxide Compounds, Organic Peroxides. Kirk-Othmer Encyclopedia of Chemical Technology. John Wiley & Sons, New York. Schaper, M., 1993. Development of a database for sensory irritants and its use in establishing occupational exposure limits. Am. Ind. Hyg. Assoc. J. 54, 488-544.

Gagnaire, F., Marignac, B., Hecht, G., Héry, M., 2002. Sensory Irritation of Acetic Acid, Hydrogen Peroxide, Peroxyacetic acid and their Mixture in Mice. Ann. Occup. Hyg. 46, 97-102.